

Cyclohexanecarboxamide

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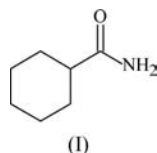
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The title compound, $C_7H_{13}NO$, forms $R_2^2(8)$ N—H \cdots O hydrogen-bonded dimers and C_4 N—H \cdots O-linked chains, which are further stabilized by a C—H \cdots O interaction. The combination of these interactions results in a hydrogen-bonded network parallel to (100), with a motif that can be described by the secondary graph set $R_6^4(16)$. The existence of the same hydrogen-bonding motif in 1-phenylcyclopentanecarboxamide and 1-(2-bromophenyl)cyclohexanecarboxamide [Lemmerer & Michael (2008). *CrystEngComm*, **10**, 95–102] indicates that replacing the H atom on position 1 with a more bulky group does not necessarily disrupt the observed hydrogen-bonding pattern. The presence of a C—H \cdots O interaction to stabilize the $R_6^4(16)$ network does, however, seem to be required. In addition, the title compound is isomorphous with a previously published structure of cyclopentanecarboxamide [Winter *et al.* (1981). *Acta Cryst.* **B37**, 2183–2185].

Comment

The hydrogen-bonding capabilities of amides have been extensively researched (Taylor *et al.*, 1984; Leiserowitz & Schmidt, 1969), and they have been exploited directly, or as precursors, in crystal engineering and the pharmaceutical industry (Reddy *et al.*, 2006), to name but a few. In addition, a systematic study of the effect of different 2,6-disubstitution on the structure of phenylamides and their consequent thermal behaviour has been published (Omondi *et al.*, 2005), as well as a similar structural study of a series of slightly more complex 1-arylcyloalkanecarboxamides (Lemmerer & Michael, 2008). In a study of symmetric and asymmetric imides and their polymorphs, the title compound, (I), was synthesized as a precursor for imide synthesis.



Compound (I) crystallizes in the space group $C2/c$ with one molecule in the asymmetric unit (Fig. 1). The molecule adopts

a chair conformation in which the amide group is rotated to be almost perpendicular to the cyclohexyl ring; atom N1 is orientated such that it lies almost eclipsed relative to atom H2, the N1—C1—C2—H2 torsion angle being -2.2° .

The structure of (I) contains two distinctive types of N—H \cdots O hydrogen bonds. One of these [N1—H1A \cdots O1ⁱⁱ; symmetry code: (ii) $-x + \frac{1}{2}, -y + \frac{3}{2}, -z + 1$; Fig. 2 and Table 1] is a hydrogen bond between two amide molecules to form an $R_2^2(8)$ dimer (Etter *et al.*, 1990; Bernstein *et al.*, 1995), with the molecules related to each other through a centre of inversion, while the other [N1—H1B \cdots O1^{iv}; symmetry code: (iv) $x, -y + 1, z + \frac{1}{2}$; Fig. 2 and Table 1] is an interaction along the c axis to form a C_4 hydrogen-bonded chain, in which the molecules are related to each other by a c -glide plane. This C_4 chain is further stabilized by a C—H \cdots O interaction (C2—H2 \cdots O1^{iv}; Fig. 2 and Table 1). The combination of the two

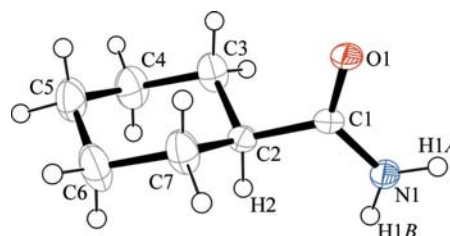


Figure 1
The molecular structure of (I), showing the atomic numbering scheme. Displacement ellipsoids are drawn at the 50% probability level.

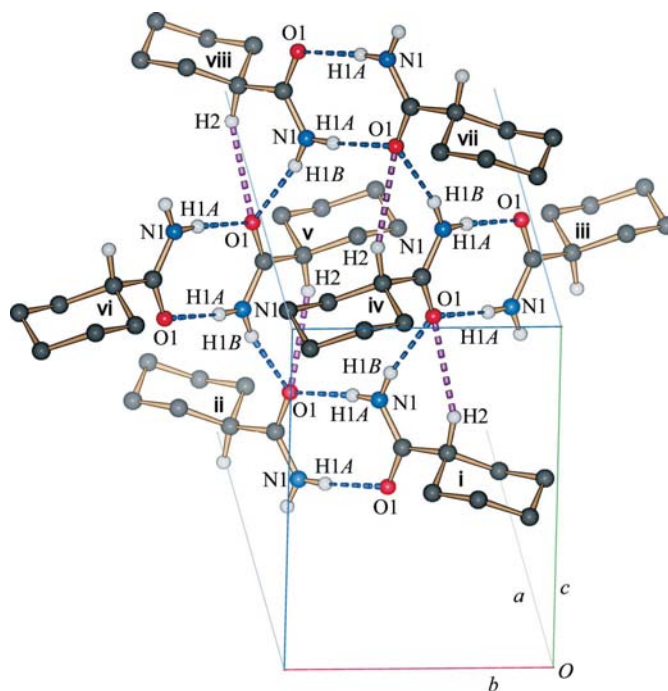


Figure 2
The intermolecular N—H \cdots O and C—H \cdots O hydrogen-bonding network (dashed lines) in the structure of (I), where molecules are connected to each other to form dimers and chains. The combination of these results in a hydrogen-bonded sheet running parallel to (100). H atoms not involved in these interactions have been omitted for clarity. [Symmetry codes: (i) x, y, z ; (ii) $-x + \frac{1}{2}, -y + \frac{3}{2}, -z + 1$; (iii) $-x + \frac{1}{2}, y - \frac{1}{2}, -z + \frac{3}{2}$; (iv) $x, -y + 1, z + \frac{1}{2}$; (v) $-x + \frac{1}{2}, y + \frac{1}{2}, -z + \frac{3}{2}$; (vi) $x, -y + 2, z + \frac{1}{2}$; (vii) $x, y, z + 1$; (viii) $-x + \frac{1}{2}, -y + \frac{3}{2}, -z + 2$.]

N—H...O interactions results in the hydrogen-bonded dimers being almost perpendicular to each other [$C1-C2 \cdots C2^{iv}-C1^{iv} = -90.99(16)^\circ$] and in the formation of a hydrogen-bonded network parallel to (100) and centred at $x = 0.25$. The combination of the two hydrogen bonds also results in a motif that can be described by the secondary graph set $R_6^4(16)$.

A search of the Cambridge Structural Database (CSD; Version 5.30, November 2008 release; Allen, 2002) led to the discovery of the isomorphous structure of cyclopentanecarboxamide (CSD refcode BARFEF; Winter *et al.*, 1981), which has an identical hydrogen-bonding pattern despite the presence of disorder in the five-membered ring. The $R_6^4(16)$ motif can also be found in the structures of 1-phenylcyclopentanecarboxamide and 1-(2-bromophenyl)cyclohexanecarboxamide, indicating that replacing atom H2 with a more bulky group does not necessarily disrupt this hydrogen-bond pattern (Lemmerer & Michael, 2008). However, a C—H...O interaction does seem to be required as it is present in all four structures. In the case of 1-phenylcyclopentanecarboxamide and 1-(2-bromophenyl)cyclohexanecarboxamide, the C—H...O interaction occurs between the amide O atom and one of the ring CH₂ groups.

Experimental

The title compound was prepared as described by Lumsden (1905). The product was recrystallized from ethanol using a slow evaporation technique at room temperature, giving a 71% yield of colourless plate-like crystals [m.p. 458–460 K (literature value 459–460.5 K; McElvain & Starn, 1955)]. ¹H NMR (300 MHz, CDCl₃, Me₄Si): δ 5.69 (1H, s, N—H), 5.52 (1H, s, N—H), 2.15 (1H, tt, $J = 3.5$ and 11.6 Hz, H1), 1.93–1.69 (5H, m, H2, H3, H4, H5, H6), 1.48–1.30 (5H, m, H2, H3, H4, H5, H6); ¹³C NMR (75 MHz, CDCl₃, Me₄Si): δ 178.7 (C6), 44.8 (C1), 297 (C2, C6), 25.7 (C3, C5), 25.6 (C4); ATR-IR: ν_{\max} 3337 (m, b, N—H), 3154 (m, b, N—H), 2927 (m, sh, C—H), 2851 (m, sh, C—H), 1635 (s, C=O), 1428 (s, C—N), 1344 (w, sh), 1285 (m, sh), 1230 (m, sh), 1154 (m, sh), 666 (s) cm⁻¹.

Crystal data

C ₇ H ₁₃ NO	$V = 1515.4(3) \text{ \AA}^3$
$M_r = 127.18$	$Z = 8$
Monoclinic, $C2/c$	Mo $K\alpha$ radiation
$a = 24.624(3) \text{ \AA}$	$\mu = 0.07 \text{ mm}^{-1}$
$b = 6.6934(9) \text{ \AA}$	$T = 173 \text{ K}$
$c = 9.4030(13) \text{ \AA}$	$0.48 \times 0.37 \times 0.10 \text{ mm}$
$\beta = 102.088(3)^\circ$	

Data collection

Bruker APEXII CCD area-detector diffractometer	1828 independent reflections
9664 measured reflections	1440 reflections with $I > 2\sigma(I)$
	$R_{\text{int}} = 0.068$

Refinement

$R[F^2 > 2\sigma(F^2)] = 0.056$	82 parameters
$wR(F^2) = 0.143$	H-atom parameters constrained
$S = 1.05$	$\Delta\rho_{\max} = 0.28 \text{ e \AA}^{-3}$
1828 reflections	$\Delta\rho_{\min} = -0.16 \text{ e \AA}^{-3}$

Table 1

Hydrogen-bond geometry (\AA , $^\circ$).

$D-H \cdots A$	$D-H$	$H \cdots A$	$D \cdots A$	$D-H \cdots A$
N1—H1A...O1 ⁱⁱ	0.88	2.06	2.9394 (16)	175
N1—H1B...O1 ^{iv}	0.88	1.99	2.8549 (15)	166
C2—H2...O1 ^{iv}	1.00	2.56	3.4283 (17)	145

Symmetry codes: (ii) $-x + \frac{1}{2}, -y + \frac{3}{2}, -z + 1$; (iv) $x, -y + 1, z + \frac{1}{2}$.

All H atoms were positioned geometrically and allowed to ride on their parent atoms [$C-H = 1.00$ (CH) or 0.99 \AA (CH₂), $N-H = 0.88 \text{ \AA}$ (NH₂) and $U_{\text{iso}}(\text{H}) = 1.2U_{\text{eq}}(\text{C}, \text{N})$].

Data collection: APEX2 (Bruker, 2005); cell refinement: APEX2; data reduction: SAINT (Bruker, 2005); program(s) used to solve structure: SHELXS97 (Sheldrick, 2008); program(s) used to refine structure: SHELXL97 (Sheldrick, 2008); molecular graphics: ORTEP-3 for Windows (Farrugia, 1997) and SCHAKAL99 (Keller, 1999); software used to prepare material for publication: WinGX (Farrugia, 1999) and PLATON (Spek, 2009).

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Supplementary data for this paper are available from the IUCr electronic archives (Reference: FG3132). Services for accessing these data are described at the back of the journal.

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